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Wyatt Tenhaeff

Milo Koretsky

Department of

Chemical

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1.2 An approximate solution can be found if we combine

Equations 1.4 and

$$1.5: \frac{1}{2} mV^2 =$$

$$kT \quad \text{molecular}$$

$$= kT \quad \text{molecular}$$

$$\frac{1}{2} mV^2 = kT$$

Assume the

temperature is 22

° C.

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found if we combine
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1.5: molecular V_m
 $\frac{1}{2} \frac{m}{kT}$ molecular
 $\frac{1}{2} \frac{m}{kT}$ ek. 3. m. kT
V. 3 Assume the
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