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Engineering Chemistry LAB VIVA Questions :-What are Hard waters? How are the waters classified based on the degree of Hardness? How is hardness of water caused? How is temporary hardness be removed? How do you express the total hardness of water? What is EDTA? Write the structural formula for EDTA. Why is disodium salt of EDTA preferred to EDTA?

107 TOP Engineering CHEMISTRY LAB VIVA Questions and ...

B.E. FIRST YEAR -ENGINEERING CHEMISTRY LAB VIVA-VOCE VOLUMETRIC ANALYSIS 1. Why hot liquids should not be taken in the Burette? A. Hot liquids should not be placed in the burette because the instrument is calibrated at a much lower temperature(15 to 20o C) 2. Which meniscus is read in case of colored solution taken in a burette? A.

B.E. FIRST YEAR ENGINEERING CHEMISTRY LAB VIVA-VOCE

VIVA VOCE Q 1; - What do u mean by alkanity of water? Ans. It is defiend as the amount of standard acid in mg required to neutralizeone litre of given water sample. Q 2 :Which constituent are responsible for alkalinity in water? Ans;-It is due to presence of hydroxide ion, carbonate ions and bicarbonate.

ENGINEERING CHEMISTRY CH-103-F LAB MANUAL I, II SEMESTER B ...

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Viva Questions with answers. Question.1.What is a standard solution ? Answer. A solution whose strength is known is called a standard solution. Question.2.What is a normal solution ? Answer. A solution containing one gram-equivalent mass of the solute per litre of the solution is called a normal solution. Question.3.What is the equivalent mass of KMnO_4 when it acts as oxidizing agent in acidic medium ? Answer.

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practical organic chemistry, we have extended the scope of the work so that it covers most of the needs of students working for an Honours or Special Degree. To meet the needs of the advanced students, preparations have now been included to illustrate, for example, reduction by lithium aluminium hydride and by the Meerwein-Ponndorf-

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800+ TOP MECHANICAL Engineering LAB VIVA Questions and Answers

INORGANIC CHEMISTRY I. COMMON DEFINITIONS Atomic number—the number of protons in the nucleus of an atom. Avogadro's number—the number of molecules

File Type PDF Engineering Chemistry Practical Viva Questions With Answers

(6.023 1023) in one gram-mole of a substance. Equilibrium constants for the reaction $aA + bB \rightleftharpoons cC + dD$ where reaction is in solution, ($[]$ refers to molarity) where reaction is in the gas phase,

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