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2.15. An aqueous solution of 2 percent non-volatile solute exerts a pressure of 1.004 bar at the boiling point of the solvent. What is the molecular mass of the solute ? Sol: According to Raoult's Law, 2.16 Heptane and octane

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form an ideal solution. At 373 K, the vapour pressures of the two liquid components are 105.2 kPa and 46.8 kPa ...

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is the best indicator about
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