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(A\* ...*

AS OCR Chemistry -  
Amount of Substance  
and Atomic Structure

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Volume = 24.0 dm<sup>3</sup>

mol<sup>-1</sup> at room

temperature and

pressure, RTP Avogadro

constant,  $N_A = 6.02 \times$

$10^{23}$  mol<sup>-1</sup> Specific

heat capacity of water,  $c$

$= 4.18$  J g<sup>-1</sup> K<sup>-1</sup> Ionic

product of water,  $K_w$

$= 1.00 \times 10^{-14}$  mol<sup>2</sup>

dm<sup>6</sup> at 298 K 1 tonne =

10<sup>6</sup> g Arrhenius

equation:  $k = A e^{-E_a/RT}$

or  $\ln k = -E$

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A H432 Data Sheet -  
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