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[CHAPTER 5 REVIEW The Periodic Law SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered \(a\) according to decreasing atomic mass. \(b\) according to Mendeleev ' s original design. \(c\) according to increasing atomic number. \(d\) based on when they were discovered.](#)

[5 The Periodic Law](#)

[The left side of the periodic table is composed of metals, which have few valence electrons. Moving to the right, there are more electrons and their ...](#)

[ANSWERS To TOPIC 5 Review Questions 12. 21. 24. 27. 30. 33. 36. 39. 42. 45. 48. 51. 54. 57. 60. 63. 66. 69. 72. 4 3 3 2 4 4 4 1 1 1 4 2 3 3 1 1 1 4 3 1. 10. 13. 16. 19. 22. 25.](#)

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[• The elements of Group 2 of the periodic table are called the alkaline-earth metals. •beryllium, magnesium, calcium, strontium, barium, and radium](#)

[Section 2 Electron Configuration and the Periodic Table Chapter 5 • Group 2 metals are less reactive than the alkali metals, but are still too reactive to be found in nature in pure form.](#)

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The Periodic Table of the Elements is useful for revealing patterns and trends in the elements. Which statement accurately describes a pattern in the size of atomic radii in the Periodic Table of the Elements?

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Chemistry End of Chapter Exercises. Using the periodic table, classify each of the following elements as a metal or a nonmetal, and then further classify each as a main-group (representative) element, transition metal, or inner transition metal: (a) uranium (b) bromine (c) strontium (d) neon (e) gold (f) americium

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