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ANSWER 1. c 2. a 3.

b 2. a. alcohol b.

water c. the gels 3.

The mixture is a colloid. The properties are consistent with those reported in Table 3 on page 404 of the text. The particle size is small, but not too small, and the mixture

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12 1. Molarity, M
Moles solute Liters
solution $M = \frac{M}{1 \text{ mol}} \frac{\text{mol}}{\text{L}}$
L = Molarity is a
standard used to
express the
concentration of a

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Review Moles solute

Liters solution $M = M$

1 mol L =

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Solids

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SECTION 3 SHORT

ANSWER Answer the

following questions in

the space provided. 1.

Describe the errors

made by the following

students in making

molar solutions. a.

James needs a 0.600

M solution of KCl.

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Answers Liquids And
Solids

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Review multiple
choice (30) define &
identify suspensions
& solutions define an
alloy define & identify
electrolytes &
nonelectrolytes
factors that affect the
rate of dissolution
definitions of

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unsaturated,
saturated &
supersaturated
solutions general
rules for predicting
whether a solute is
soluble in a solvent
definition of solubility
effects of temperature
& pressure on the
solubility of gases and
solids in liquids
definitions of molarity
(M) and molality (m)

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Problem: FS show all
show all steps. Match
each of the key terms
above to the definition
that best fits it. ...

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SECTION 1 SHORT

ANSWER Answer the following questions in the space provided. 1.

Match the type of mixture on the left to its representative particle diameter on the right. c solutions (a) larger than 1000 nm a suspensions (b) 1 nm to 1000 nm b colloids (c) smaller than 1 nm 2. Identify

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the solvent in each of
the following
examples:

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. SHORT ANSWER

Answer the following
questions in the
space provided. 1.

Describe the errors

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made by the following students in making molar solutions. 3.

James needs a 0.600 M solution of KCl. He measures out 0.600 g of KCl and adds 1 L of water to the solid.

James made several errors.

CHAPTER REVIEW

Solutions

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Modern Chemistry

Chapter 12-Solutions

Section 1- Types of

Mixtures Solutions are

homogeneous

mixtures of two or

more substances in a

single phase. Soluble

describes a

substance as capable

of being dissolved.

Solvent is the

dissolving medium in

a solution. Solute is

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the substance that is dissolved in a solution.

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Mixtures Solutions are
homogeneous

mixtures of two or

more substances in a
single phase. Soluble

describes a

substance as capable
of being dissolved.

Solvent is the

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dissolving medium in a solution. Solute is the substance that is dissolved in a solution.

Answers

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Answers

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Review: Solutions
Although there are

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this set (28) 2 or more

substances mixed,
not chemically

combined. Mixture.

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Type of solutions:
does not display
Tyndall effect, will
separate by filtering.
Homogenous
Solution. Type of
solutions: ionic ...

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ANSWER 1. c 2. a... h

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