

## Chapter 12 Organic Chemistry Wade

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Chapter 12 ©2010, Prentice Hall Organic Chemistry, 7th Edition L. G. Wade, Jr. Infrared Spectroscopy and Mass Spectrometry 2. Chapter 12 2 Introduction • Spectroscopy is a technique used to determine the structure of a compound. • Most techniques are nondestructive (it destroys little or no sample).

12 - Infrared Spectroscopy and Mass Spectrometry - Wade 7th

Organic Chemistry, 7e (Wade) Chapter 12 Infrared Spectroscopy and Magnetic Spectrometry 1) Which bond in the structure below would give rise to the highest absorption frequency in an IR spectrum? A) sp<sup>2</sup> C-H B) sp<sup>3</sup> C-H C) C-D D) C=C Answer: A Diff: 2 Section: 12.2 2) Absorption \_\_\_\_\_ is the measurement of the amount of light absorbed by a ...

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Organic molecules must have N-H or O-H to be able to form a hydrogen bond. iii. The bond is polar and the H becomes positively charged( Electrophile- likes electrons). iv.

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and Sulfides 1) What is the hybridization of the oxygen atom in dialkyl ethers? A)  $sp^3$  B)  $sp^2$  C)  $sp$  D)  $s$  E)  $p$  Answer: A Diff: 1 Section: 14.2 2) Which of the following corresponds to the COC bond angle in dimethyl ether?

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